General Chemistry I – CHEM1411
Department of Physical & Environmental Sciences
Summer I 2017

A. COURSE INFORMATION
   Course number/section: CHEM1411.001
   Class meeting time: MTWR 1:00-2:55PM
   Class location: CI-122
   Course Website: https://bb9.tamucc.edu/

B. INSTRUCTOR INFORMATION
   Instructor: Jim Silliman, PhD
   Office location: Harte Research Institute 120 (HRI-120)
   Office hours: MTWR 4:00-5:15PM
   Telephone: 361-825-3718
   e-mail: james.silliman@tamucc.edu
   Appointments: Email or call to schedule an appointment outside office hours

C. COURSE DESCRIPTION
   Catalog Course Description
   The foundation course in chemistry. Stoichiometry, chemical equilibria, atomic structure, chemical bonding, periodic properties, thermodynamics, chemical kinetics, and descriptive chemistry of the elements. Laboratory involves development of basic skills. This course counts toward the natural science component of the University Core Curriculum. Either CHEM 1305 - Introductory Chemistry or CHEM 1411, but not both, may be applied towards the core requirement.

D. PREREQUISITES AND COREQUISITES
   Prerequisites
   None.

   Corequisites
   SMTE0093

E. REQUIRED TEXTBOOK(S), READINGS AND SUPPLIES
   Recommended Textbook
   Texas A&M University-Corpus Christi (Chemistry, The Molecular Nature of Matter and Change), by Silberberg/Amateis, 2015, Seventh Ed.
   * Online access code not necessary.

   Optional Textbook(s) or Other References
   None.
F. STUDENT LEARNING OUTCOMES AND ASSESSMENT

Assessment is a process used by instructors to help improve learning. Assessment is essential for effective learning because it provides feedback to both students and instructors. A critical step in this process is making clear the course’s student learning outcomes that describe what students are expected to learn to be successful in the course. The student learning outcomes for this course are listed below. By collecting data and sharing it with students on how well they are accomplishing these learning outcomes, students can more efficiently and effectively focus their learning efforts. This information can also help instructors identify challenging areas for students and adjust their teaching approach to facilitate learning.

By the end of this course, students should be able to:
1. Name ionic and molecular compounds.
2. Understand periodic behavior of the elements.
4. Carry out stoichiometric calculations.

G. INSTRUCTIONAL METHODS AND ACTIVITIES

Students will have a chance to learn from each other as they participate in in-class group assignments. Group assignments will complement the latest material covered in lecture and will challenge their grasp of general chemistry concepts.

H. MAJOR COURSE REQUIREMENTS AND GRADING

The average of 4 exam grades (including the final) and quizzes will determine the lecture grade. There will be regular homework assignments from each chapter during the semester. You must do the homework problems in order to perform well on exams and quizzes. Four 100-point exams, usually covering 2-3 chapters, are planned (including the final exam). Final letter grades for the lecture course will be assigned as follows: A: 90%, B: 80%, C: 70%, D: 60%, F: < 60%. When justified by the exam statistics and item analysis data, “curve” points may sometimes be added to every student’s final course average, “curving” grades up. Grades will never be “curved” down.

Course Grade - 25% of the general chemistry grade is from general chemistry lab. At the end of the semester, after completing all the lab assignments and exams, you will be assigned a lab grade by your lab instructor. All complaints and concerns about the lab grade should be directed to the lab instructor. You’ll receive more information about lab in the lab syllabus.

<table>
<thead>
<tr>
<th>Lecture Grade</th>
<th>Lecture % X 0.75</th>
<th>75</th>
</tr>
</thead>
<tbody>
<tr>
<td>Lab Grade</td>
<td>Lab % X 0.25</td>
<td>25</td>
</tr>
<tr>
<td>Course Grade</td>
<td></td>
<td>100</td>
</tr>
<tr>
<td>ACTIVITY</td>
<td>Points</td>
<td></td>
</tr>
<tr>
<td>---------------------</td>
<td>--------</td>
<td></td>
</tr>
<tr>
<td>3 Exams</td>
<td>300</td>
<td></td>
</tr>
<tr>
<td>Final Exam</td>
<td>100</td>
<td></td>
</tr>
<tr>
<td>Quizzes</td>
<td>40</td>
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<tr>
<td><strong>Total</strong></td>
<td><strong>440</strong></td>
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</table>

I. COURSE CONTENT/SCHEDULE

Tentative Schedule:

<table>
<thead>
<tr>
<th>DATE</th>
<th>TOPIC</th>
<th>CHAPTER</th>
<th>EXAM</th>
</tr>
</thead>
<tbody>
<tr>
<td>Tues. 5/30/17</td>
<td>Keys to Chemistry</td>
<td>1.1, 1.3-1.6</td>
<td></td>
</tr>
<tr>
<td>Wed. 5/31/17</td>
<td>Keys to Chem./Comps. Of Mat.</td>
<td>1.3-1.6/2.1-2.6</td>
<td></td>
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<tr>
<td>Thurs. 6/01/17</td>
<td>Comps. of Mat./Stoich./Quantum T.</td>
<td>2.1-2.6, 3.1, 7</td>
<td></td>
</tr>
<tr>
<td>Mon. 6/05/17</td>
<td>Quantum T. &amp; Atomic Str.</td>
<td>7</td>
<td>Exam 1</td>
</tr>
<tr>
<td>Tues. 6/06/17</td>
<td>Quantum T./Electron Config.</td>
<td>7/8</td>
<td></td>
</tr>
<tr>
<td>Wed. 6/07/17</td>
<td>Electron Config.</td>
<td>8</td>
<td></td>
</tr>
<tr>
<td>Thurs. 6/08/17</td>
<td>Ch. Bonding/Comps. of Mat.</td>
<td>8</td>
<td></td>
</tr>
<tr>
<td>Mon. 6/12/17</td>
<td>Energy &amp; Chem. Reactions</td>
<td>9.1-9.3/2.8</td>
<td>Exam 2</td>
</tr>
<tr>
<td>Tues. 6/13/17</td>
<td>Shapes of Molecules</td>
<td>9.5/10</td>
<td></td>
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<tr>
<td>Wed. 6/14/17</td>
<td>Shapes of Molecules/Cov. Bonding</td>
<td>10/11</td>
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<tr>
<td>Thurs. 6/15/17</td>
<td>Cov. Bonding/Stoichiometry</td>
<td>11/3.1</td>
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<tr>
<td>Mon. 6/19/17</td>
<td>Stoichiometry</td>
<td>3.1-3.5</td>
<td>Exam 3</td>
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<tr>
<td>Tues. 6/20/17</td>
<td>Chem. Reactions</td>
<td>4</td>
<td></td>
</tr>
<tr>
<td>Wed. 6/21/17</td>
<td>Chem. Reactions/Gases</td>
<td>4/5</td>
<td></td>
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<tr>
<td>Thurs. 6/22/17</td>
<td>Gases/Thermochemistry</td>
<td>5/6</td>
<td></td>
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<tr>
<td>Mon. 6/26/17</td>
<td>Thermochemistry</td>
<td>6</td>
<td></td>
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<tr>
<td>Tues. 6/27/17</td>
<td>Chem. Bonding</td>
<td>9.2&amp;9.4</td>
<td></td>
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<tr>
<td>Wed. 6/28/17</td>
<td>Review</td>
<td></td>
<td></td>
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<tr>
<td>Thurs. 6/29/17</td>
<td>Final Exam</td>
<td>1-11</td>
<td>Final Ex.</td>
</tr>
</tbody>
</table>

Exam Schedule:
- Exam 1: Monday, 6/05/17 (1:00-2:00)
- Exam 2: Monday, 6/12/17 (1:00-2:00)
- Exam 3: Monday, 6/19/17 (1:00-2:00)
- Final Exam: Thursday, 6/29/17 (1:00-2:55)

Note: Changes in this course schedule may be necessary and will be announced to the class by the Instructor. The assignments and exams shown are directly related to the Student Learning Outcomes described in Section F.
J. COURSE POLICIES

Attendance/Tardiness
The student is expected to arrive on time prepared to take notes and work on in-class problems with pen or pencil, paper, calculator and colored markers/pencils. If absent, it is the responsibility of the student to obtain missed information from a classmate. Missed information includes not only lecture notes, but also any possible information regarding syllabus changes.

Make-up Exams
There will be no make-up exams or quizzes for this class. If you miss one lecture exam, your final exam grade will be counted twice to replace the missed exam. This applies to ONE exam only. If you miss more than one, you will receive a zero for the additional missed exam(s). For those students who do not miss an exam, your final exam grade will be counted twice to replace your lowest exam grade (assuming that this improves your overall grade). Do not show up late to an exam, no student will be admitted to the exam after the first exam-taker has left.

Extra Credit
All extra credit is included on each exam and there will be an extra credit quiz offered during the semester.

Cell Phone Use
Before you enter the lecture hall turn OFF your cell phone! Beepers must also be turned off or put on silent mode. Electronic interruptions will NOT be tolerated! Cell phones must be put away and stored out of sight during all quizzes and exams or you will receive a zero!!

Laptop Use
You are welcome to use a laptop to take notes during class. Do not use it to check email, facebook, youtube videos, etc. These other uses are considered a distraction and you will be asked to leave.

Food in Class
Drinks and snacks are allowed. Do not bring in a meal – this is not a cafeteria.

Participation
You are expected to be attentive and participate in asking/answering questions and also in group assignments.

K. COLLEGE AND UNIVERSITY POLICIES

• Academic Integrity (University)
   It is expected that university students will demonstrate a high level of maturity, self-direction, and ability to manage their own affairs. Students are viewed as individuals who possess the qualities of worth, dignity, and the capacity for self-direction in
• **Classroom/Professional Behavior**

Texas A&M University-Corpus Christi, as an academic community, requires that each individual respect the needs of others to study and learn in a peaceful atmosphere. Under Article III of the Student Code of Conduct, classroom behavior that interferes with either (a) the instructor’s ability to conduct the class or (b) the ability of other students to profit from the instructional program may be considered a breach of the peace and is subject to disciplinary sanction outlined in article VII of the Student Code of Conduct. Students engaging in unacceptable behavior may be instructed to leave the classroom. This prohibition applies to all instructional forums, including classrooms, electronic classrooms, labs, discussion groups, field trips, etc.

• **Deadline for Dropping a Course with a Grade of W (University)**

The grade of W will be assigned to any student officially dropping a course. Please consult with the instructor before you decide to drop to be sure it is the best thing to do. Just stopping attendance and participation WILL NOT automatically result in your being dropped from the class. Should dropping the course be the best course of action, visit the Office of the University Registrar for the Course Drop Form that must be submitted. No student is eligible to receive a W without completing the official drop process by this deadline. Please consult the Academic Calendar ([http://www.tamucc.edu/academics/calendar/](http://www.tamucc.edu/academics/calendar/)) for the last day to drop a course.

• **Grade Appeals (College of Science and Engineering)**

As stated in University Procedure 13.02.99.C2.01, Student Grade Appeal Procedures, a student who believes that he or she has not been held to appropriate academic standards as outlined in the class syllabus, equitable evaluation procedures, or appropriate grading, may appeal the final grade given in the course. The burden of proof is upon the student to demonstrate the appropriateness of the appeal. A student with a complaint about a grade is encouraged to first discuss the matter with the instructor. For complete details, including the responsibilities of the parties involved in the process and the number of days allowed for completing the steps in the process, see University Procedure 13.02.99.C2.01, Student Grade Appeal Procedures. These documents are accessible through the University Rules website at [http://www.tamucc.edu/provost/university_rules/index.html](http://www.tamucc.edu/provost/university_rules/index.html), and the College of Science and Engineering Grade Appeals webpage at [http://sci.tamucc.edu/students/GradeAppeal.html](http://sci.tamucc.edu/students/GradeAppeal.html). For assistance and/or guidance in the grade appeal process, students may contact the chair or director of the appropriate department or school, the Office of the College of Science and Engineering Dean, or the Office of the Provost.

• **Disability Services**

The Americans with Disabilities Act (ADA) is a federal anti-discrimination statute that provides comprehensive civil rights protection for persons with disabilities. Among other
things, this legislation requires that all students with disabilities be guaranteed a learning environment that provides for reasonable accommodation of their disabilities. If you believe you have a disability requiring an accommodation, please call (361) 825-5816 or visit Disability Services in Corpus Christi Hall 116.

If you are a returning veteran and are experiencing cognitive and/or physical access issues in the classroom or on campus, please contact the Disability Services office for assistance at (361) 825-5816.

http://disabilityservices.tamucc.edu/

L. **OTHER INFORMATION**

- **Tutoring and Test Taking Strategies**
  To be successful in this course, and most others, you must develop good note-taking skills, organization skills, study habits, and test-taking strategies from the very beginning. Your instructor, seminar leaders and TA’s are always available for help, but don’t wait until it’s too late! It is important that you are aware that the Center for Academic Student Achievement in Room 216 of the library provides free tutoring, test-taking strategies, and extra help. Take advantage of this service! Should you have test anxiety, stress problems, or need help with study skills, the University Counseling Center (Driftwood 107: 825-2703) also provides a free service.

- **Use of Electronic Devices During Exams**
  Any use of an electronic device (palm pilot, Cell Phone, MP3 player, CD player, computer …) during an exam is strictly prohibited. Any use of such a device will be considered an attempt to cheat on the exam and will result in a 0 on the exam although more severe actions may be considered. Calculators may be allowed on exams when needed, but only for mathematical operations. The use of programmable calculators to store or retrieve information during an exam will be considered an attempt to cheat on the exam. Also, if a calculator is discovered to have saved programs or information that could be used as an unfair advantage on the exam, this will be considered an attempt to cheat on the exam. Programs or operators that aid in mathematical operations such as a quadratic equation calculator may be used.

- **Assigned Homework**
  Homework assignments for this class are not graded. They are designed to complement class lectures and to prepare you for exams and quizzes …DO NOT GET BEHIND. Complete the homework problems before each exam.

- **Academic Advising**
  The College of Science & Engineering requires that students meet with an Academic
Advisor as soon as they are ready to declare a major. The Academic Advisor will set up a degree plan, which must be signed by the student, a faculty mentor, and the department chair. The College's Academic Advising Center can be accessed at: http://www.sci.tamucc.edu/advising/index.html

- **Statement of Academic Continuity**
  In the event of an unforeseen adverse event, such as a major hurricane and classes could not be held on the campus of Texas A&M University–Corpus Christi; this course would continue through the use of Blackboard and/or email. In addition, the syllabus and class activities may be modified to allow continuation of the course. Ideally, University facilities (i.e., emails, web sites, and Blackboard) will be operational within two days of the closing of the physical campus. However, students need to make certain that the course instructor has a primary and a secondary means of contacting each student.

**GENERAL DISCLAIMER**

I reserve the right to modify the information, schedule, assignments, deadlines, and course policies in this syllabus if and when necessary. I will announce such changes in a timely manner during regularly scheduled lecture periods.